

# Heavy Metal and Metalloid Contamination From the Battle of Verdun

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# Preview

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- ▶ Background and contaminated area
- ▶ Water pollution
- ▶ Contaminant mobility
- ▶ Arsenic contamination



# Background

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- ▶ Battle of Verdun: Feb. 21 – Dec. 18, 1916
- ▶ High heavy metal contamination
  - ▶ Lead and Copper
    - ▶ Shell casings
  - ▶ Zinc
    - ▶ Fuses
  - ▶ Arsenic
    - ▶ Organoarsenic compounds (chemical weapons)
  - ▶ Many other contaminants
- ▶ Contaminated area of 96 km<sup>2</sup>



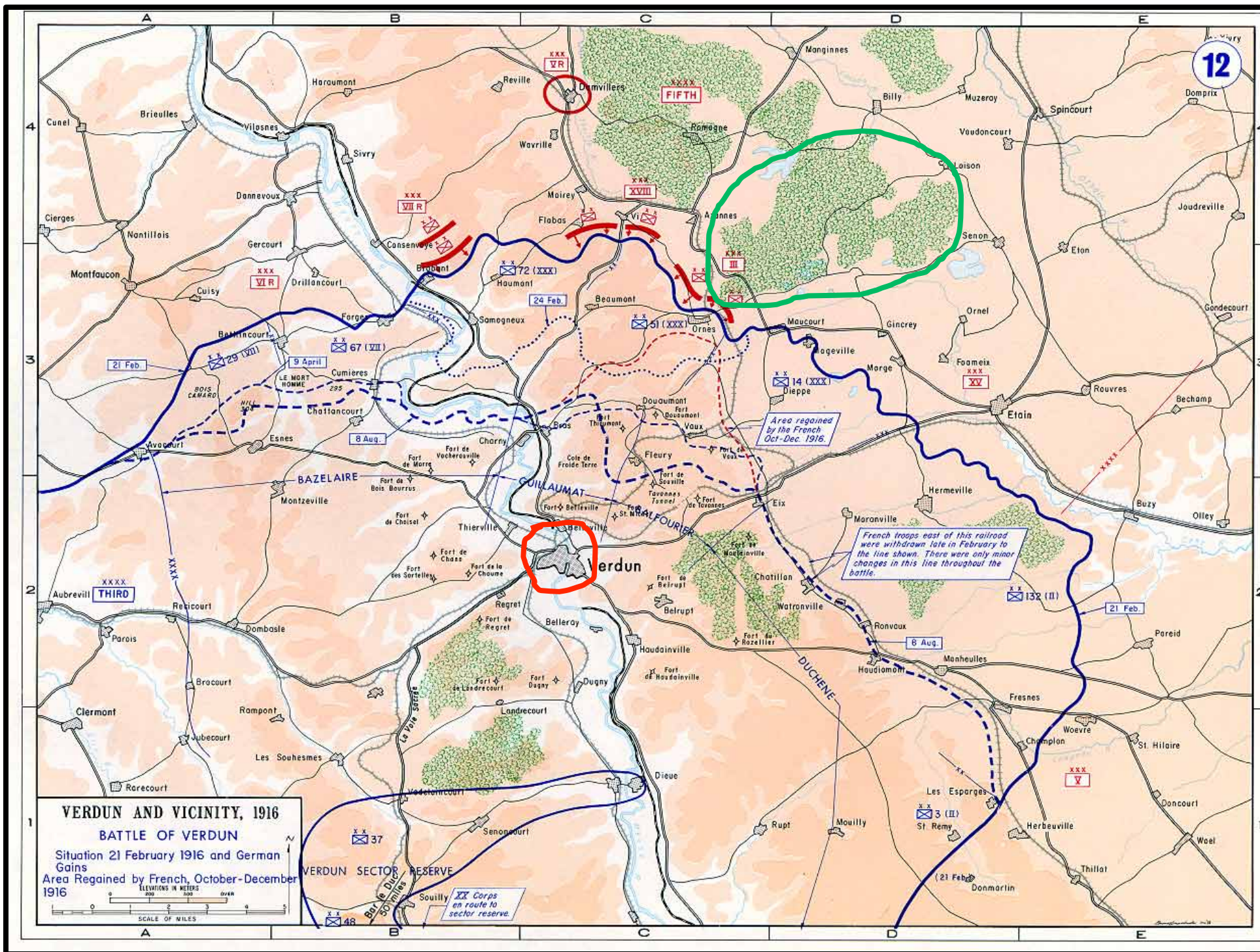
# Contaminated Area

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- ▶ **Most contaminated site of the battle**
  - ▶ Place-à-Gaz in the Spincourt Forest
  - ▶ 20 km Northeast of Verdun, France
- ▶ **Cause of contamination**
  - ▶ Previous fighting
  - ▶ Burning of leftover weapons
    - ▶ 1928
- ▶ **Massive amounts of arsenic**



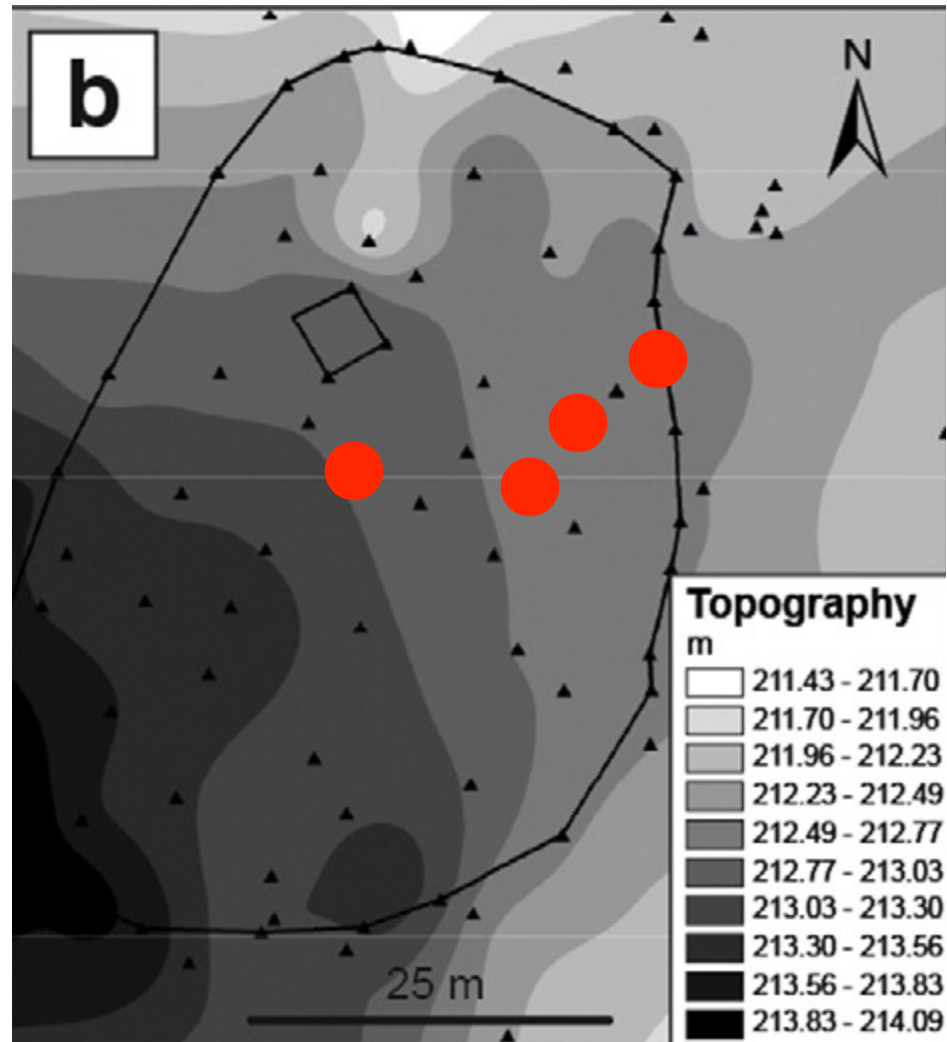






## Contaminated Area (cont.)

- ▶ Average rainfall: 758 mm
- ▶ Average Temp: 10.7°C
- ▶ Little plant life



# Water Pollution

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- ▶ **Leaching test**
  - ▶ Copper, lead, zinc
  - ▶ Concentrations determined by AAS
- ▶ **Percolation test**
  - ▶ Arsenic
  - ▶ Oxidation of As(III) to As(V)
  - ▶ Concentrations determined by ion exchange and AAS

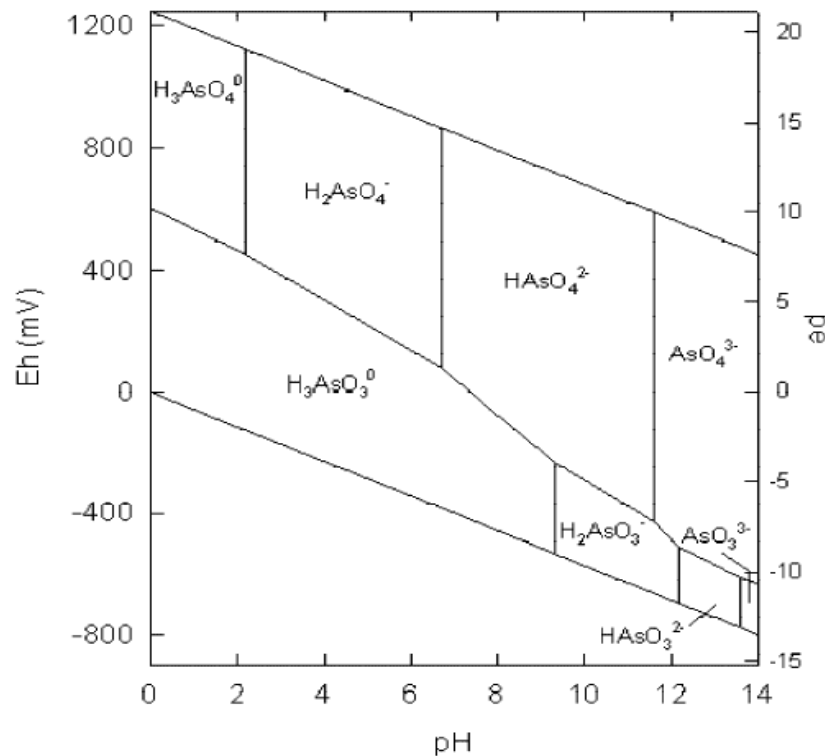
Sample	As	Cu	Pb	Zn
1	77.2	3.0	2.9	103.9
2	40.5	2.8	4.6	70.8
3	21.8	2.8	4.2	59.9
4	26.3	3.0	2.2	74.5
WHO*	0.01	2.0	0.01	3.0

\*World Health Organization water quality standards for clean drinking water



# Arsenic Speciation

- ▶ Oxidation of As(III) to As(V)
- ▶ Ion exchange separation
  - ▶ As(III) is uncharged while As(V) is charged at pH 5-6
- ▶ 2% As(III) and 98% As(V)





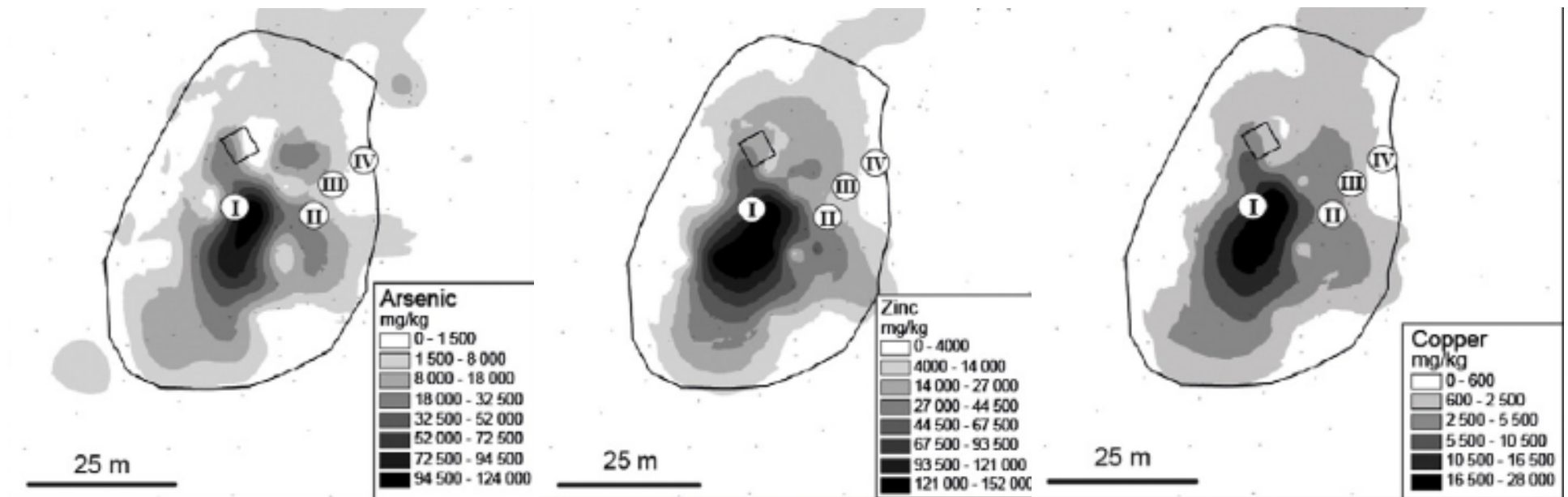
# Contaminant Mobility

- ▶ Copper and Lead

- ▶ Constant concentration over each sample

- ▶ Arsenic and Zinc

- ▶ High conc. in Sample I decreasing through to Sample 4



# Arsenic Contamination

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- ▶ Arsenic adsorption to Hydrous Ferric Oxides (HFO)
  - ▶  $\text{Fe}_3\text{O}_4$ , Magnetite
  - ▶  $\text{FeO}(\text{OH})$ , Goethite
  - ▶  $\text{Fe}_2\text{O}_3 \cdot 0.5\text{H}_2\text{O}$ , Ferrihydrite
- ▶ pH values of 5 and 8
- ▶ Temperature values of  $10^\circ\text{C}$  and  $25^\circ\text{C}$

As	Ca	Cl	Cu	F	Fe	K	Mg	Na	P	Pb	Zn
7.67	0.83	1.16	0.30	0.05	3.89	0.26	0.29	0.85	0.72	0.31	10.3



# Mineral Saturation

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10°C

pH	Magnetite	Goethite	Ferrihydrite
5	-4.83	-3.69	-6.51
6	3.61	-0.70	-3.51
7	11.41	2.30	-0.52
8	19.08	5.28	2.46

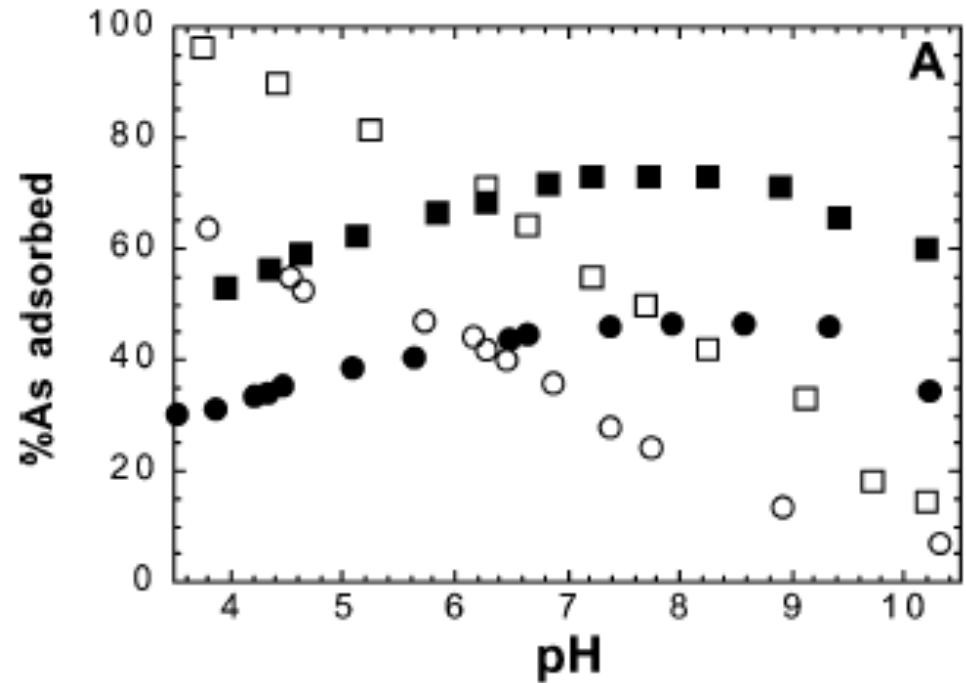
25°C

pH	Magnetite	Goethite	Ferrihydrite
5	-2.11	-2.74	-5.44
6	5.88	0.26	-2.44
7	13.84	3.25	0.55
8	21.71	6.20	3.50



# Arsenic Adsorption

- ▶ As(III) neutral at pH 5-6
  - ▶ Adsorbs better at higher pH
- ▶ As(V) charged at pH 5-6
  - ▶ Adsorbs better at lower pH
- ▶ As(III) and As(V) adsorb at similar rates at pH 6
- ▶ Higher pH limits movement of arsenic



Boxes = 100 μM

Circles = 50 μM

Solid = As(III)

Open = As(V)

# Summary

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- ▶ Verdun Battlefield is highly contaminated
- ▶ Water contamination
- ▶ Arsenic and Zinc runoff
- ▶ Higher pH causes less mobility of Arsenic





# References

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Thouin, H.; Forsteir, L. L.; Gautret, P. Battaglia-Brunet, F. Characterization and mobility of arsenic and heavy metals in soils polluted by the destruction of arsenic-containing shells from the Great War. Science of the total environment. 550. (2016) 658-669.

Ngai, K. K. Arsenic Speciation and Evaluation Of An Adsorption Media In Rupandehi And Nawalparasi Districts Of Nepal. MIT. June 2002.

Dixit, S.; Hering, J. Comparison of Arsenic(V) and Arsenic(III) Sorption onto Iron Oxide Minerals: Implications for Arsenic Mobility. Environ. Sci. Technol. 2003, 37, 4182-4189

World Health Organization. Guidelines for Drinking-water Quality. 3rd edition. Volume I. 2008.

