PHYS 360 MODERN PHYSICS II

**PLEASE READ “COVID-19 RELATED INFORMATION”**

**BASIC INFORMATION**
Course prefix, catalog number, and title: PHYS 360, 14870, Modern Physics II  
Number of credits: 3 credit hours  
Term and year: Fall 2022  
Classes: Mon, Wed, Fri 11:00 - 11:50 AM, South Eng. 221 (lecture; Mon & Wed) and South Eng. 312 (Lab; Fri)

Instructor's name: Prof. Yongki Choi  
Email Address: yongki.choi@ndsu.edu  
Office location: South Engineering 220A  
Office hours: Mon, Wed, Fri 11:50 - 1:00 pm and by appointment

**BULLETIN DESCRIPTION**
Continuation of modern physics covering molecular structure, solid state physics, nuclear and particle physics with an embedded modern physics laboratory with experiment such as atomic and molecular spectroscopy, electron diffraction, nuclear spectroscopy, photoelectric effect and computer simulation of experiments. Pre-requisite: PHYS 350

**COURSE OBJECTIVES**
The main objective of the course is to develop the conceptual and quantitative methods that are critical for a working knowledge of modern physics. The student will be able to explain modern physics concepts and to use laboratory equipment to reproduce experiments in modern physics, as well as measure physics properties described by modern physics concepts.

**REQUIRED STUDENT RESOURCES**
Recommended book: *Modern Physics* by Paul Tipler and Ralph Llewellyn  
Recommended book: *Modern Physics for Scientists and Engineers* by John Morrison

**SYLLABUS ON WEB PAGES**
Syllabus, Announcements, and Notes will be posted on our Blackboard course homepage: [https://bb.ndsu.nodak.edu](https://bb.ndsu.nodak.edu)

**HOMEWORK ASSIGNMENTS**
Homework will be posted on our Blackboard course homepage. All homework assignments are due on the dates specified.  
*Late submission will not receive credit.*

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<th>Topic</th>
<th>Reading /Assignment</th>
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<td>1</td>
<td>Course/lab introduction/preparation</td>
<td>Review lab manuals</td>
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<td>2</td>
<td>Experiments/Measurement/Data analysis</td>
<td>Review lab manuals</td>
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<td>3</td>
<td>Molecular/Atomic Spectra</td>
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<td>Solid State Physics, <strong>Midterm Exam</strong></td>
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PHYS 360: EVALUATION PROCEDURES AND GRADING CRITERIA

Final letter grades for the course will be computed using the following weights:

- Homework Assignments 20%
- Midterm Exam 20%
- Final Exam 20%
- 6 Labs Assignments 40%
- Total Points 100%

NO MAKE-UP EXAMS ARE ALLOWED

Grades: A: ≥ 85 %, B: 70 to < 85 %, C: 60 to < 70 %, D: 50 to < 60 %, F: < 50 %

Requirements and assessment of the lab reports are described in the attached document.

ATTENDANCE EXPECTATIONS

According to NDSU Policy 333 (www.ndsu.edu/fileadmin/policy/333.pdf), attendance in classes is expected. Veterans and student service members with special circumstances or who are activated are encouraged to notify the instructor as soon as possible and are encouraged to provide Activation Orders.

AMERICANS WITH DISABILITIES ACT FOR STUDENTS WITH SPECIAL NEEDS

Any students with disabilities or other special needs, who need special accommodations in this course, are invited to share these concerns or requests with the instructor and contact the Disability Services Office (www.ndsu.edu/disabilityservices) as soon as possible.

ACADEMIC HONESTY

The academic community is operated on the basis of honesty, integrity, and fair play. NDSU Policy 335: Code of Academic Responsibility and Conduct applies to cases in which cheating, plagiarism, or other academic misconduct have occurred in an instructional context. Students found guilty of academic misconduct are subject to penalties, up to and possibly including suspension and/or expulsion. Student academic misconduct records are maintained by the Office of Registration and Records. Informational resources about academic honesty for students and instructional staff members can be found at www.ndsu.edu/academichonesty.

The instructor reserves the right to adjust or modify this syllabus if it is deemed beneficial to student learning.
PHYS 360  MODERN PHYSICS LABORATORY

GENERAL INFORMATION
Each student will individually complete two experiments during the semester. Each laboratory will nominally take hours to complete. Students will rotate through the experiments during the course of the semester. Each laboratory period will replace one lecture. Additional times for lab work will be coordinated at the beginning of the semester since the experiments will not usually be completed in one hour. The laboratories will complement the sections covered in lecture and in general the material covered will not be repeated in lecture. Corresponding reading sections will be assigned in the text for the experiment.

TENTATIVE LABORATORY CHOICES

Lab Number & Topic
1. Atomic and molecular spectroscopy
2. Electron diffraction
3. Michelson interferometer
4. \( e/m \), determination of \( h \)
5. Photoelectric effect
6. Millikan oil drop experiment

LAB REPORT PREPARATION

1. Read the experiment manual before coming to lab.
2. Data is to be defended before writing up the lab. This means presenting your data to the instructor, and proving that it is adequate to meet the goals of the lab.
3. Lab reports will be prepared using Microsoft Word.
4. Images, plots, etc. will be prepared in MatLab, Excel, or your favorite software and inserted into the Word document. MatLab and Excel are available on the SE312 computers.
5. Lab reports will be submitted using BlackBoard Assignment.

CONTENTS OF LAB REPORTS
Parts A through I below must be included in all reports. One basic standard is that you must include enough information so that another student would be able to easily repeat the experiment, avoiding any problems that you experienced.

A. In an introductory section, explain the purpose of the experiment and the physics behind the experiment. In the body of your report indicate the purpose of each set of measurements or calculations you report. Clearly explain what you have done. For calculations explain what you are calculating and how it was calculated (including appropriate equations). Carefully identify all the variables used in equations and calculations. SHOW INTERMEDIATE STEPS IN ALL ANALYSIS.
B. Discuss any problems encountered in the experiment and how you overcame them.
C. Draw or copy a diagram of the experimental apparatus used to perform the experiment. Clearly show how any parts are connected. Also give a complete list of parts (pieces of equipment etc.) used in experiment.
D. Collect and record at least two sets of data for every measurement you take. Assign experimental errors to your measured data. For example, if you take a reading from an analog meter or a meter stick, an estimate of the experimental error would be some reasonable fraction of the smallest division on the scale. For digital instruments you can usually use changes in the signal over time to estimate an error.
E. Calculate errors in the physical constants or other parameters you determined in your experiment. Assume that these are random uncorrelated errors. Calculation of random errors will be discussed in class.
F. Neatly tabulate and plot your data using Excel, Matlab or other software.
G. Fitting of equations to experimental data. In experiments you are requested to fit equations to your data to determine significant experimental parameters.
H. Always plot "best-fit" functions (that is the appropriate equation using the best-fit parameter value or values) as a solid line (without points). On the same graph plot your data points as large symbols like diamonds not connected by a line. This allows a direct comparison that often tells you if something has gone wrong with your attempt to fit an equation to your data or your measurements are corrupted in some way. As discussed in class, data points should distribute evenly on both sides of your best-fit curve in a “good” fit.
I. Finally summarize your experiment. The results found and the conclusions reached should be discussed. For example, if you have determined a physical constant, one part of your summary should be to compare your value with the accepted value for this constant. A discussion of the difference in these quantities in terms of your calculated error in the constant should also be given. Discuss other problems which may produce errors.
COVID-19 RELATED INFORMATION

HEALTH AND SAFETY EXPECTATIONS
NDSU web resource for information on COVID-19: https://www.ndsu.edu/admission/fall_2020_prelim_plan

While masks are not required as we begin the 2022 fall semester, NDSU administration has determined that faculty may request mask use in their classroom. In this class, I may ask that you wear a mask to help protect my health and the health of your peers.

Where possible, please spread out within the classroom, including not sitting in the first row of the classroom, to maximize social distancing.

ATTENDANCE EXPECTATIONS
Please do not come to class
• if you are feeling ill, particularly if you are experiencing COVID-19 symptoms, or
• if you are infected, during your five-day isolation period.

You will still need to complete the assignments, exams, reading, etc. necessary to meet class learning objectives. Please contact me ASAP.

If you were exposed to COVID-19, please follow CDC guidance available below.
https://www.cdc.gov/coronavirus/2019-ncov/your-health/if-you-were-exposed.html

If you tested positive for COVID-19, please follow CDC guidance available below.

Free testing kits can be picked up at the NDSU Bookstore, Library or Student Health Service. Rapid and PCR testing is available at the Student Health Service by appointment Monday through Friday during regular business hours for both symptomatic and asymptomatic students.

If public health conditions and directives from NDSU administration change, I will let you know in writing the expectations for our class moving forward.