

# EFFECTIVE DELIVERY OF IRON NANOPARTICLES USING AMPHIPHILIC POLYSILOXANE GRAFT COPOLYMERIC VEHICLES

**Sita Krajangpan**

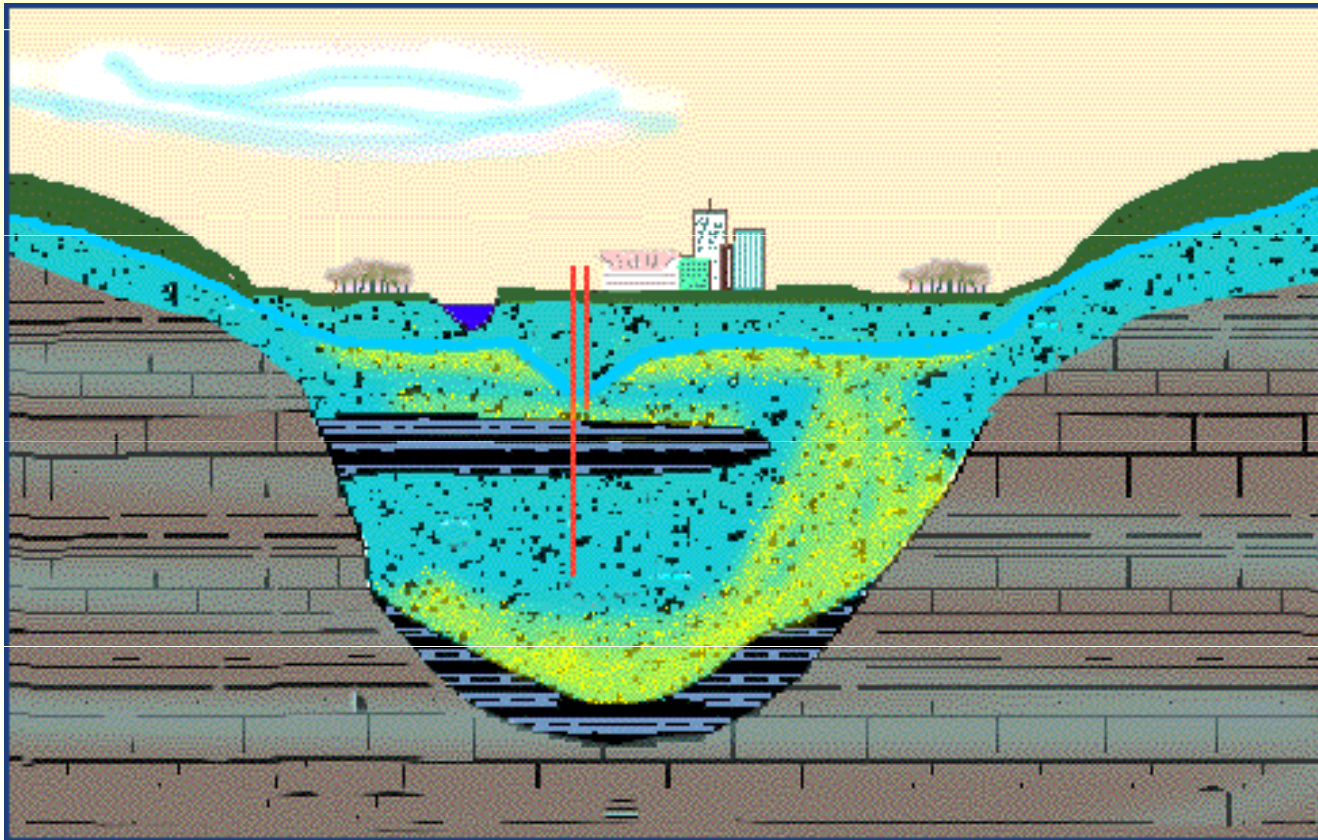
Civil Engineering Department  
North Dakota State University

# OUTLINE

- Background
- Objectives
- Hypothesis
- Research Approach
- Results
- Summary

# BACKGROUND

## Groundwater Contamination



<http://www.gwconsortium.org/>

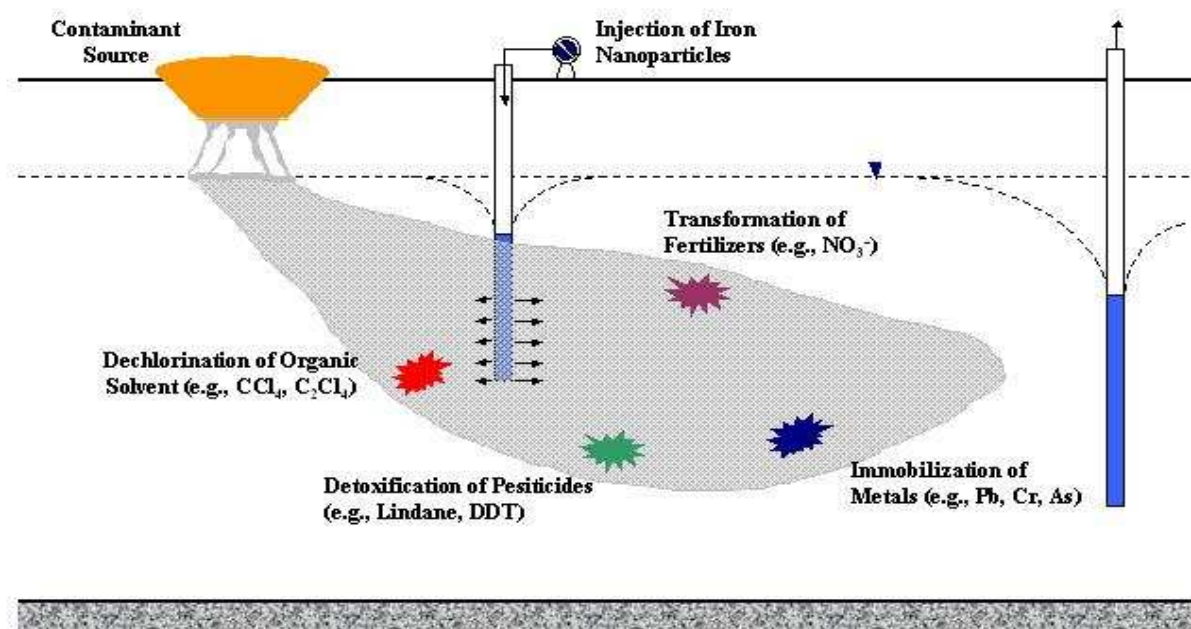
# BACKGROUND

## Groundwater Remediation

- Biological treatment
  - Inexpensive
  - Takes long time
- Physical/Chemical treatment
  - Very fast technique
  - Slightly expensive

# BACKGROUND

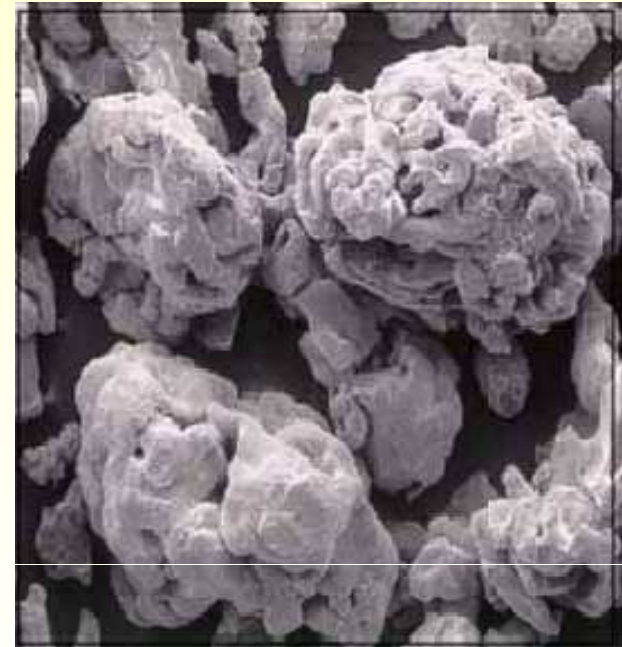
## Injection of Zero-valent Iron Nanoparticles (nZVI)



# BACKGROUND

## Why Iron Nanoparticles?

- Highly effective on several kinds of reducible contaminants
- Easy in dispersion to all spaces available (subsurface injection )
- High reactive surface area ( $\sim 35 \text{ m}^2/\text{g}$ )
- Extremely high reaction rate
- Low temperature reaction
- Economics (relatively inexpensive)



<http://enviro.nfesc.navy.mil/>

# BACKGROUND

## Treatable with Iron Nonoparticles

### ■ Methanes:

Carbon tetrachloride (CT)  
Chloroform  
Bromoform

### ■ Ethanes:

Hexachloroethane  
1,1,1-Trichloroethane (TCA)  
1,1,2,2-Trichloroethane  
1,1,2,2-Tetrachloroethane  
1,1,1,2-Tetrachloroethane  
1,1-Dichloroethane (DCA)

### ■ Ethenes:

Tetrachloroethene (PCE)  
Trichloroethene (TCE)  
1,1-Dichloroethene  
*trans*-1,2-Dichloroethene  
*cis*-1,2-Dichloroethene  
  
Vinyl chloride

### ■ Other Organics:

1,1,2-Trichlorotrifluoroethane (Freon 113)  
Trichlorofluoromethane (Freon 11)  
1,2,3-Trichloropropane  
1,2-Dichloropropane  
1,2-Dibromo-3-chloropropane  
1,2-Dibromoethane  
*n*-Nitrosodimethylamine (NDMA)  
Nitrobenzene

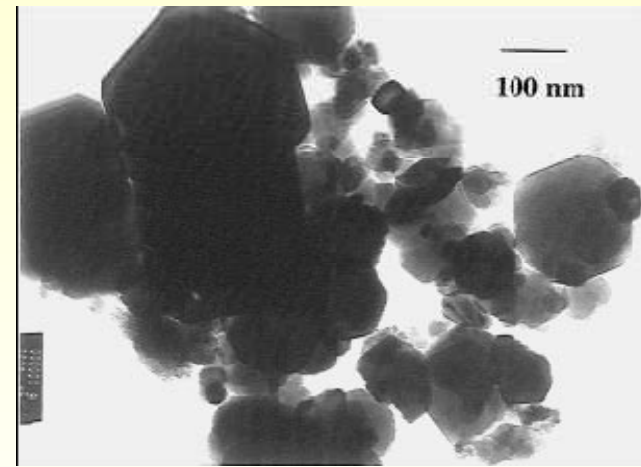
### ■ Inorganics:

Chromium, nickel  
Lead, Copper, Zinc  
Nitrate , Arsenic

# BACKGROUND

- **Zhang, et al. (1997)**

*Nanoscale particles can quickly and completely dechlorinate several chlorinated aliphatic compounds and mixture of polychlorinated biphenyls (PCB) at relatively low metal to solution ratio*



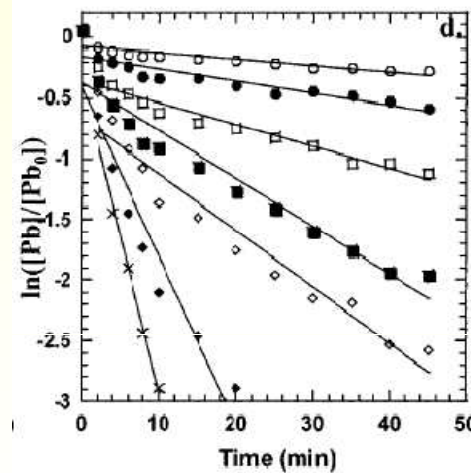
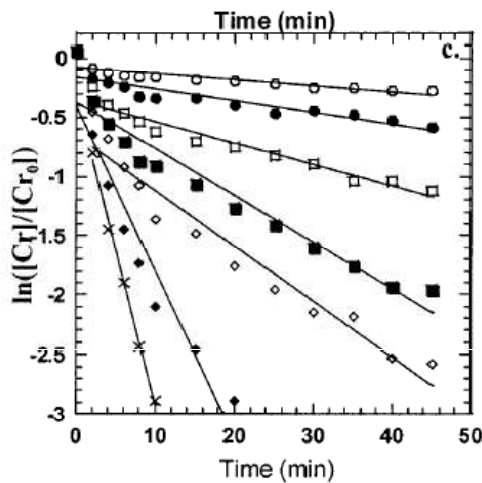
Transmission electron microscopy image of nanoscale Fe particles.



# BACKGROUND

## ■ Ponder, *et al.*(2000)

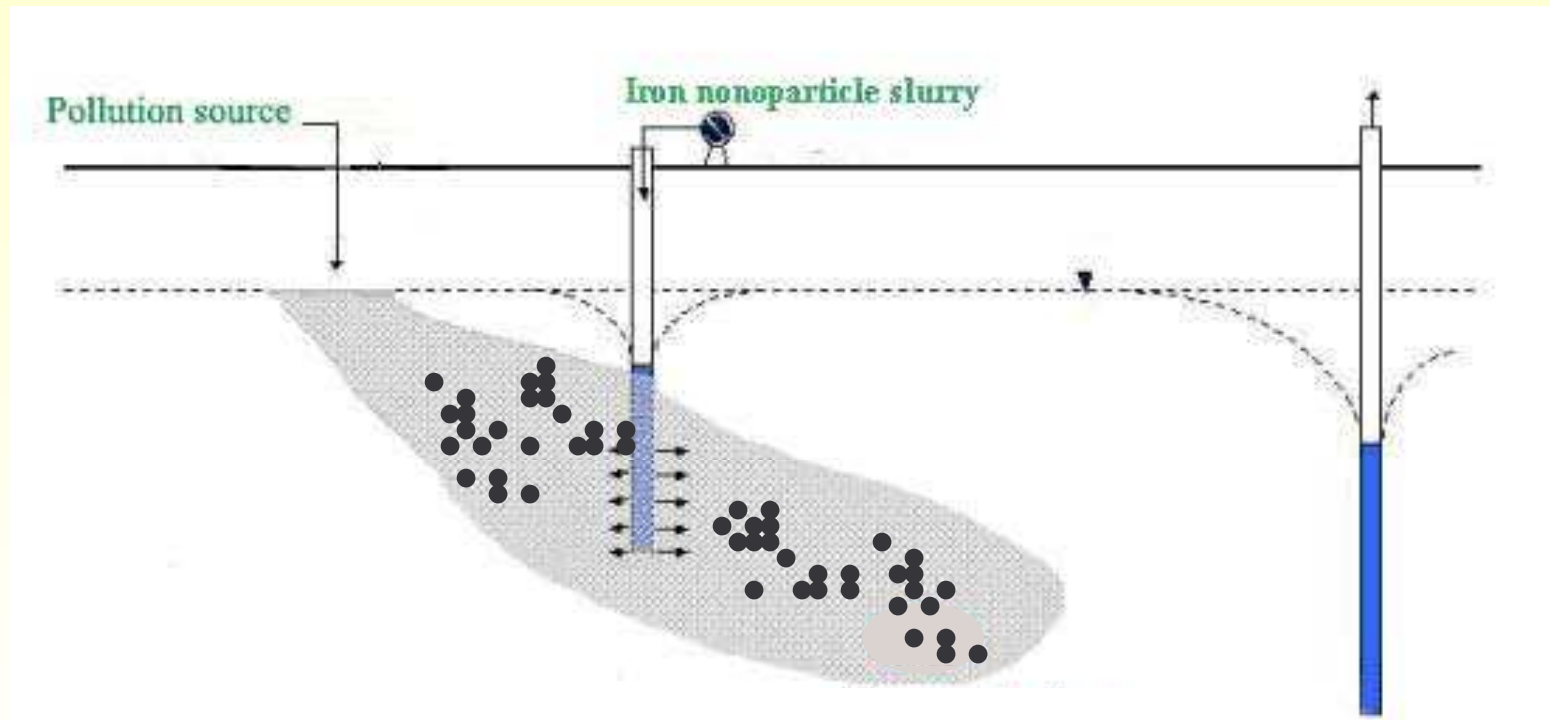
*nZVI rapidly immobilize Cr(VI) and Pb(II) in aqueous solution*



Comparison of first-order kinetics for reduction of Cr(VI) and Pb(II)

# BACKGROUND

## In-Situ Remediation using nZVI



Modified from <http://www.nsf.gov/od/lpa/news/03/pr0394.htm>

Problems: Oxidation Rate  $\uparrow$  and Dispersibility  $\downarrow$

# BACKGROUND

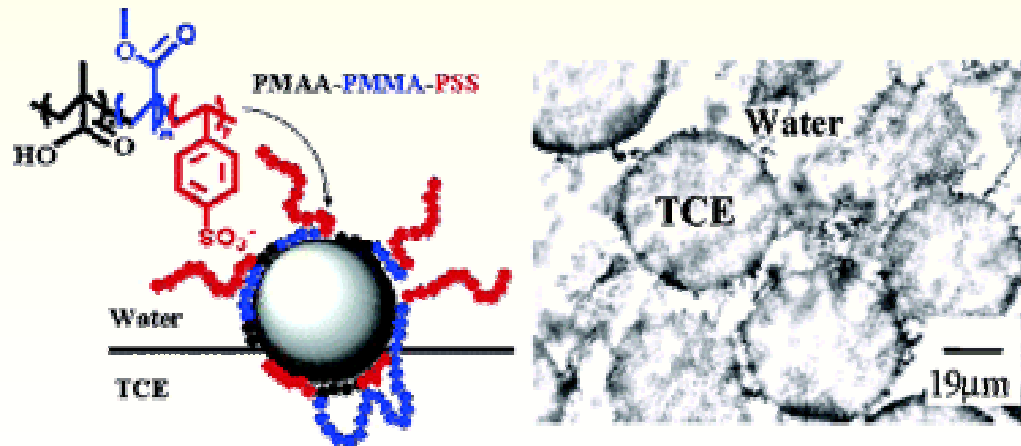
## In-situ application for nZVI (state-of-the-art)

- Injection of a carbon source needed to
  - render the site anoxic
  - prevent oxidation of  $\text{Fe}^0$  particles by non-targets
- High financial investment
- Time consuming

# BACKGROUND

- **Saleh, et al. (2005)**

*Used novel triblock copolymers to modify iron nanoparticle surfaces for promoting colloidal stability in aqueous suspension and drive into the oil/water interface*



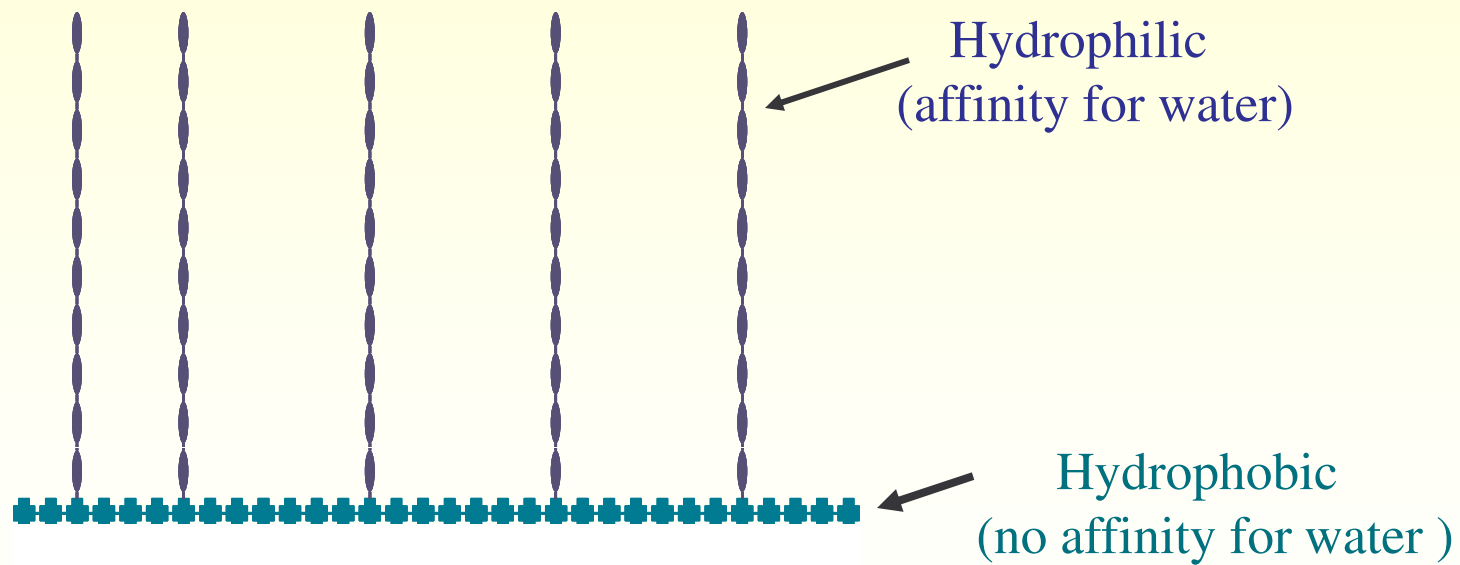
# BACKGROUND

## Type of Copolymers

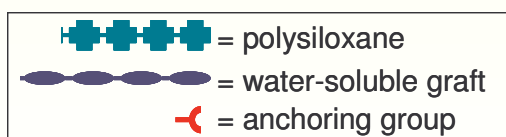
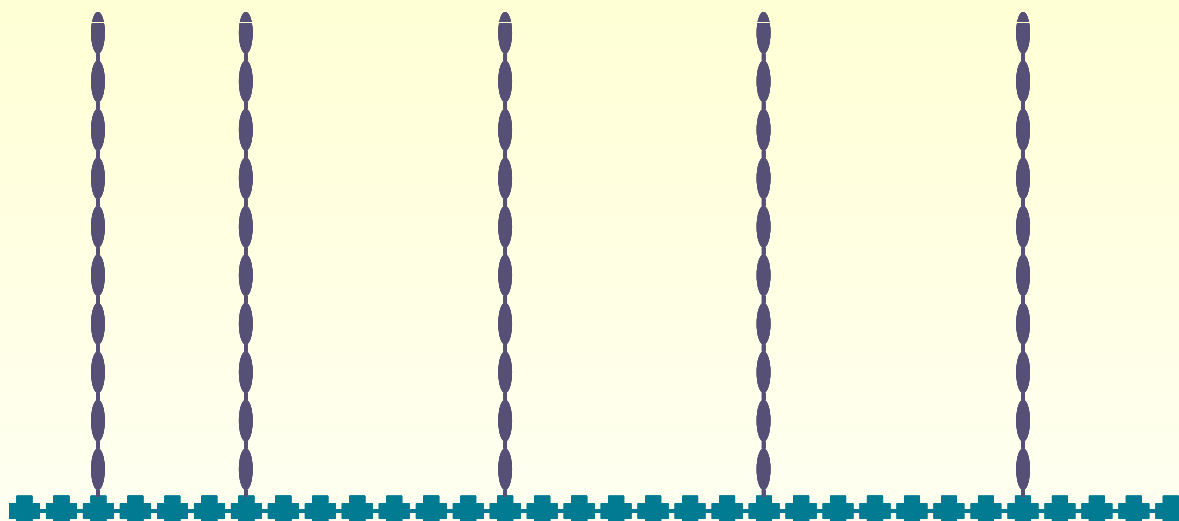
- Random Copolymers : A-A-B-B-A-A-A-B-A-B-A-A-B-B-B
- Alternating Copolymers: A-B-A-B-A-B-A-B-A-B-A-B-A-B
- Block Copolymers: A-A-A-A-B-B-B-B-A-A-A-A-B-B-B...
- Graft Copolymers: A-A-A-A-A-A-A-A-A-A-A-A-A-A-A  
                          |      |          |  
                         B      B          B  
                         |      |          |  
                         B      B          B  
                         |      |          |  
                         B      B          B  
                         |      |          |  
                         B      B          B

# BACKGROUND

## ■ Amphiphilic Graft Copolymer

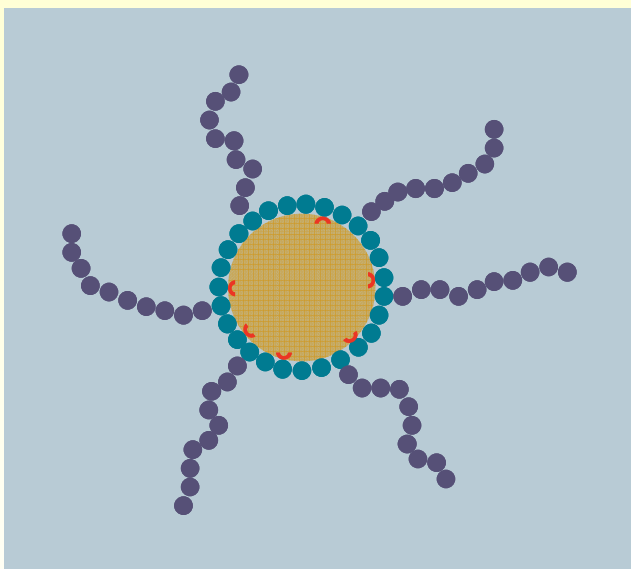


# BACKGROUND



A schematic representation of amphiphilic polysiloxane graft copolymer (APGC)

# BACKGROUND

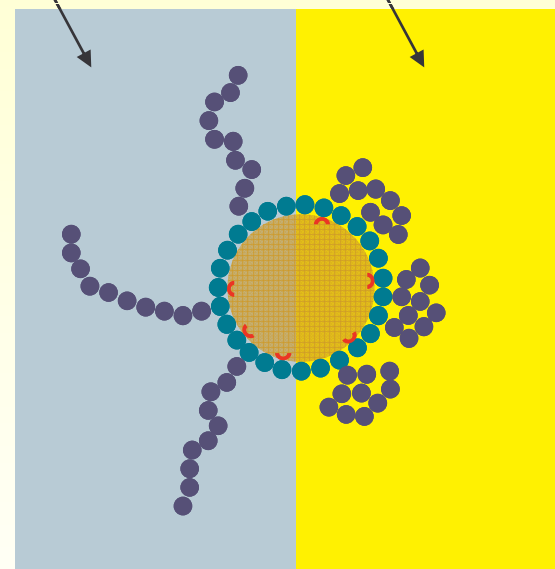


Coated Fe<sup>0</sup> Nanoparticle  
Dispersed in the Aqueous Phase



Aqueous  
Phase

Organic  
Phase



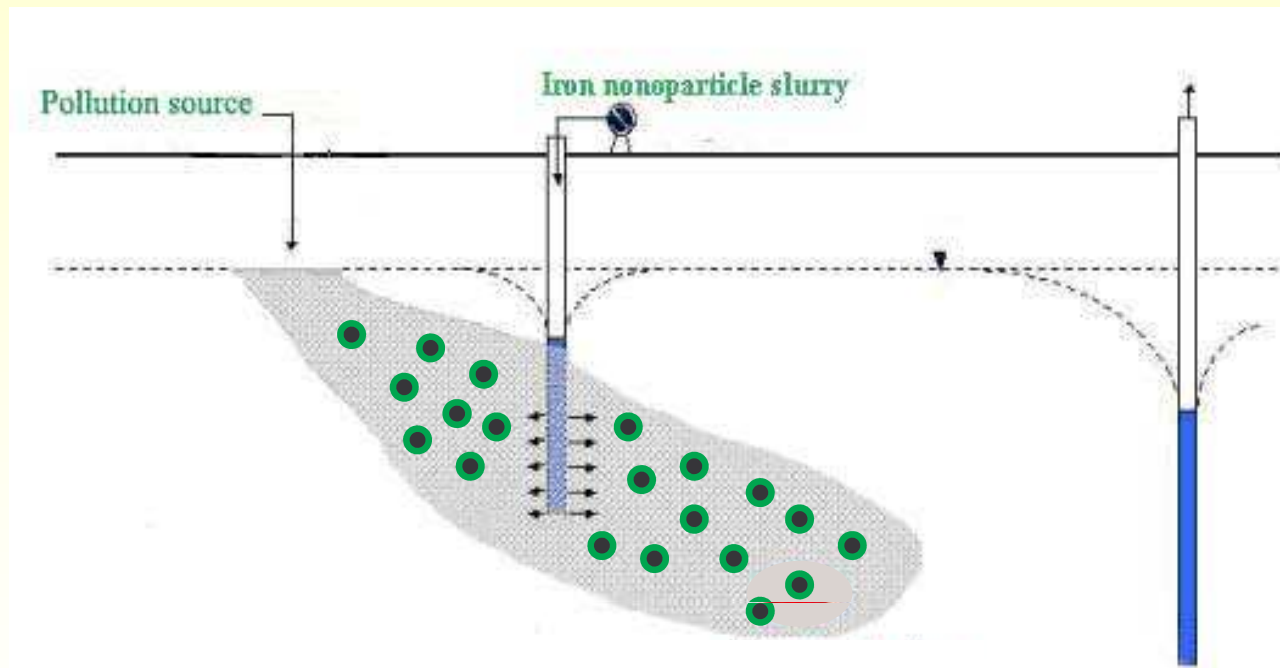
Coated Fe<sup>0</sup> Nanoparticle  
at the Aqueous/Organic Interface

A schematic representation of the polymer coated nanoparticles both in water and at the water/contaminant interface.



# BACKGROUND

## In-Situ Remediation using polymer coated iron nanoparticles slurry



Modified from <http://www.nsf.gov/od/lpa/news/03/pr0394.htm>

Oxidation Rate ↓ and Dispersibility ↑

# MAIN OBJECTIVE

Modify the Iron Nanoparticle surface using amphiphilic polysiloxane graft copolymer (APGC) to be effective for groundwater remediation

# SPECIFIC OBJECTIVES

- Synthesis and characterization of APGCs
- Synthesis and characteristic of the iron nanoparticles
- Measurement of sedimentation rate and interface affinity of polymer coated iron nanoparticles
- Comparison of reaction kinetics during polymer coated and uncoated of iron nanoparticles mediated degradation of TCE (Trichloroethylene)
- Degradation of TCE by-products and corrosion of iron nanoparticles

# HYPOTHESIS

Amphiphilic polysiloxane graft copolymers will provide the colloidal stability, protection against hydrolytic oxidation, and affinity for the water/contaminant interface needed to improve capabilities of nZVI for groundwater remediation

# RESEARCH APPROACH

- Task I : Synthesis and characterization of amphiphilic polysiloxane graft copolymers (APGC)
- Task II: Synthesis and characterization of nZVI
- Task III : High throughput measurement of sedimentation rate and interface affinity
- Task IV: Batch studies to compare the effectiveness of the polymer coated and bare nZVI for TCE degradation
- Task V: By-products of TCE and characterization of nZVI

# TASK I : Synthesis/Analysis of APGC



*Nuclear Magnetic Resonance  
Spectroscopy (NMR)*



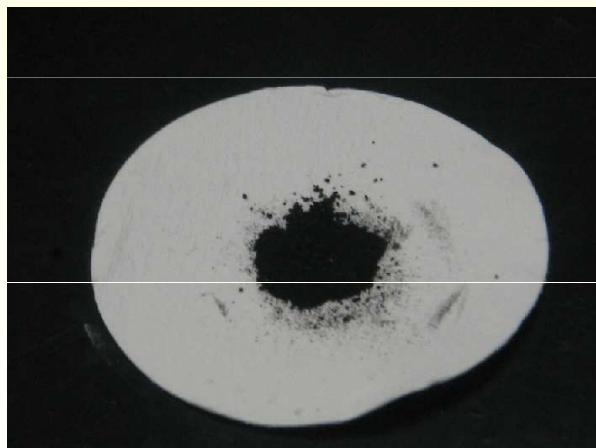
*Fourier Transform Infra Red  
spectroscopy (FTIR)*

(chemical structure of polymers)

# TASK II : Synthesis of nZVI

Using method by Liu, *et al.* (2005)

- Sodium borohydride is the key reductant
- $\text{NaBH}_4$  is added to  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$
- $\text{N}_2$  environment



*Iron nanoparticles*



*Glove Box*

## TASK II : Analysis of nZVI



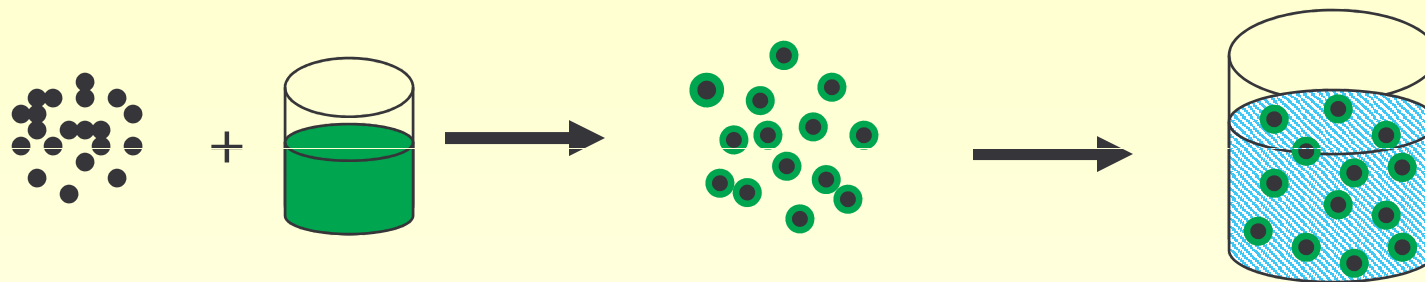
*X-ray diffraction (XRD)*  
**(composition of the nanoparticles)**



*Transmission electron microscope (TEM)*  
**(size distribution of the nanoparticles)**



# TASK III: Colloidal Study



*UV Spectrophotometer*

# TASK IV : Kinetic Studies

- Contaminant: TCE
- Batch test using polymer coated and bare nZVI
  - 20 mL vials used as a batch reactor
  - Control studies done



*Gas Chromatography-Mass Spectrometry (GC-MS)*

# TASK V : TCE by-products and characterization of Fe<sup>0</sup>

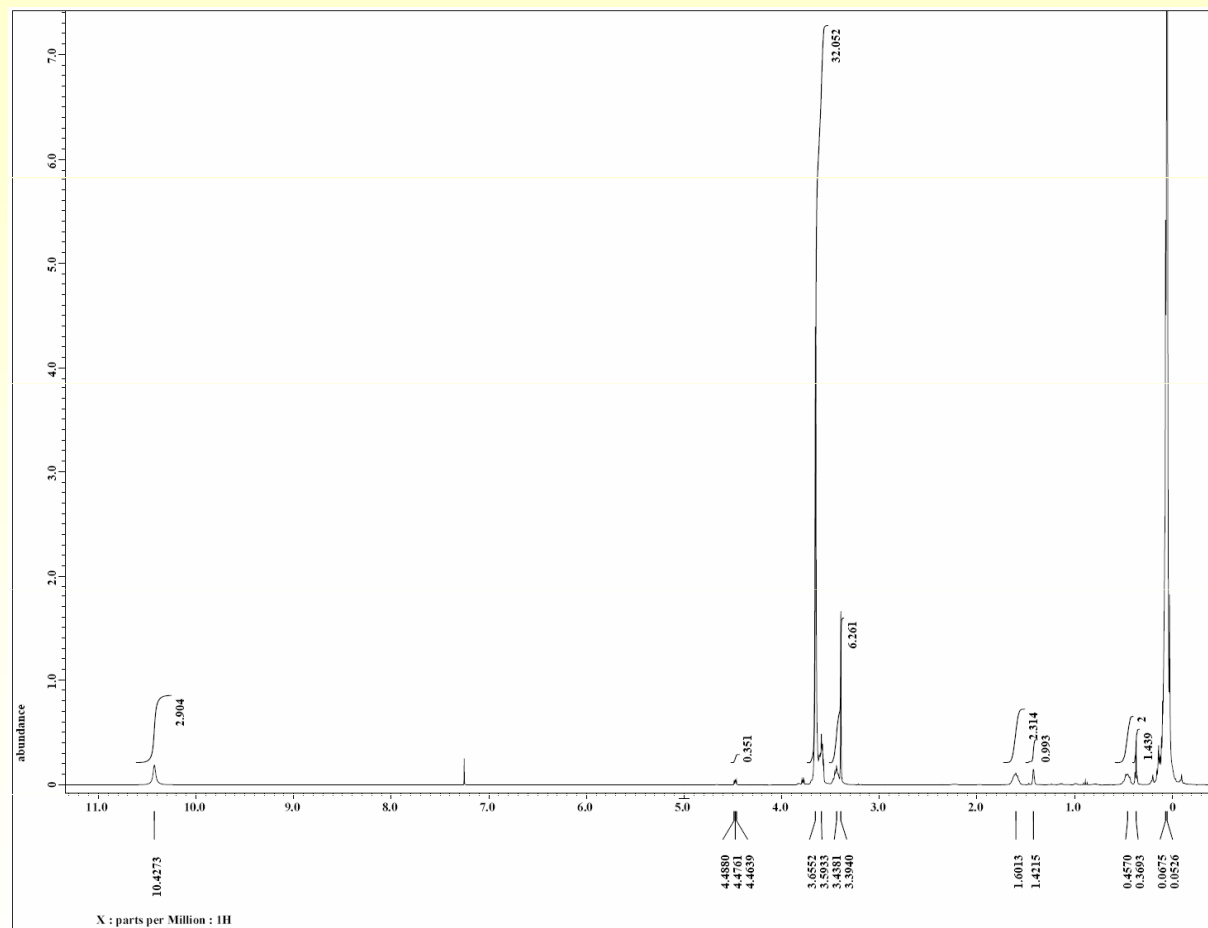


*Gas Chromatography-Mass  
Spectrometer (GC-MS)*



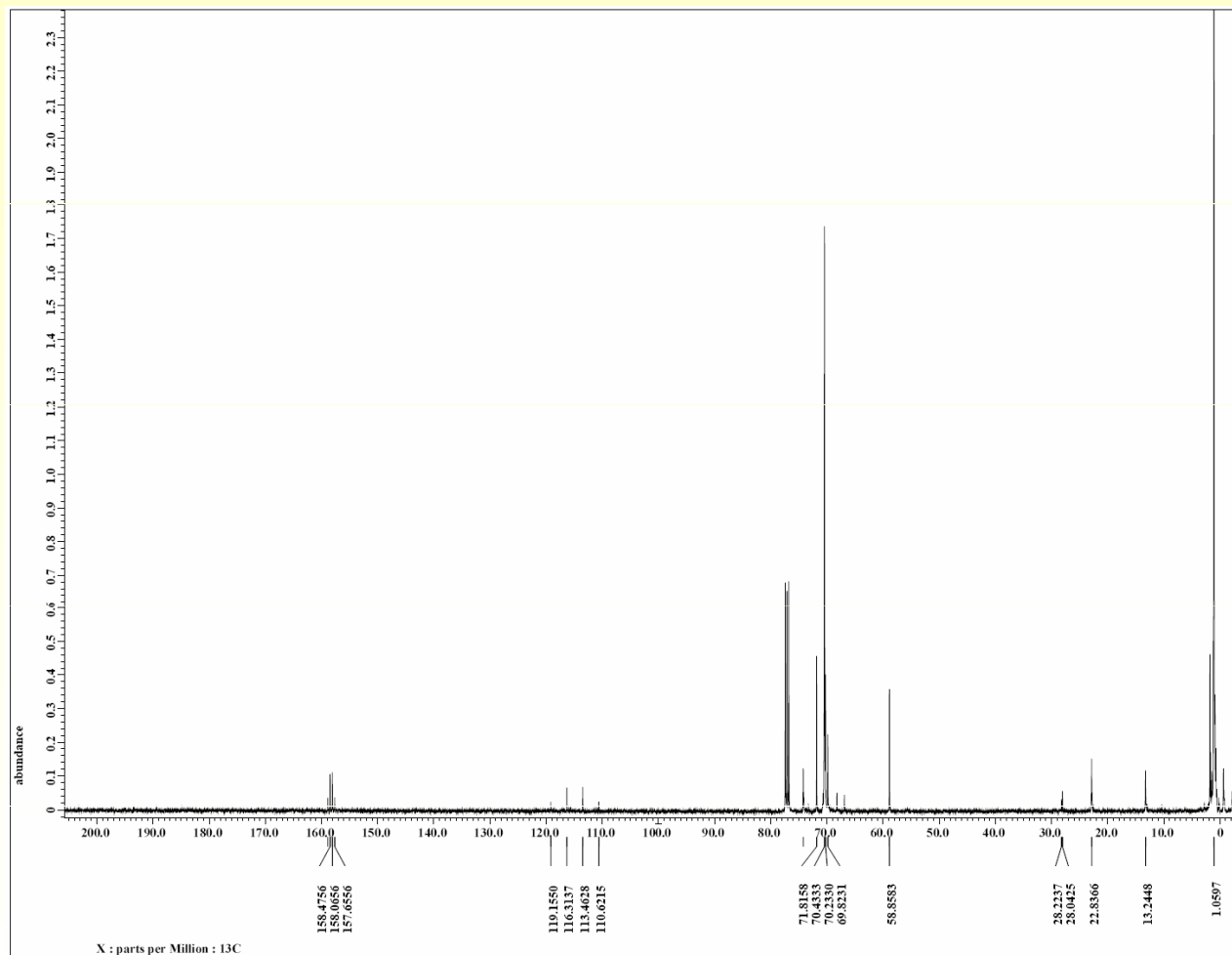
*Scanning Electron Microscopy  
(SEM) / EDX*

# RESULTS ...Task I : Synthesis of APGC



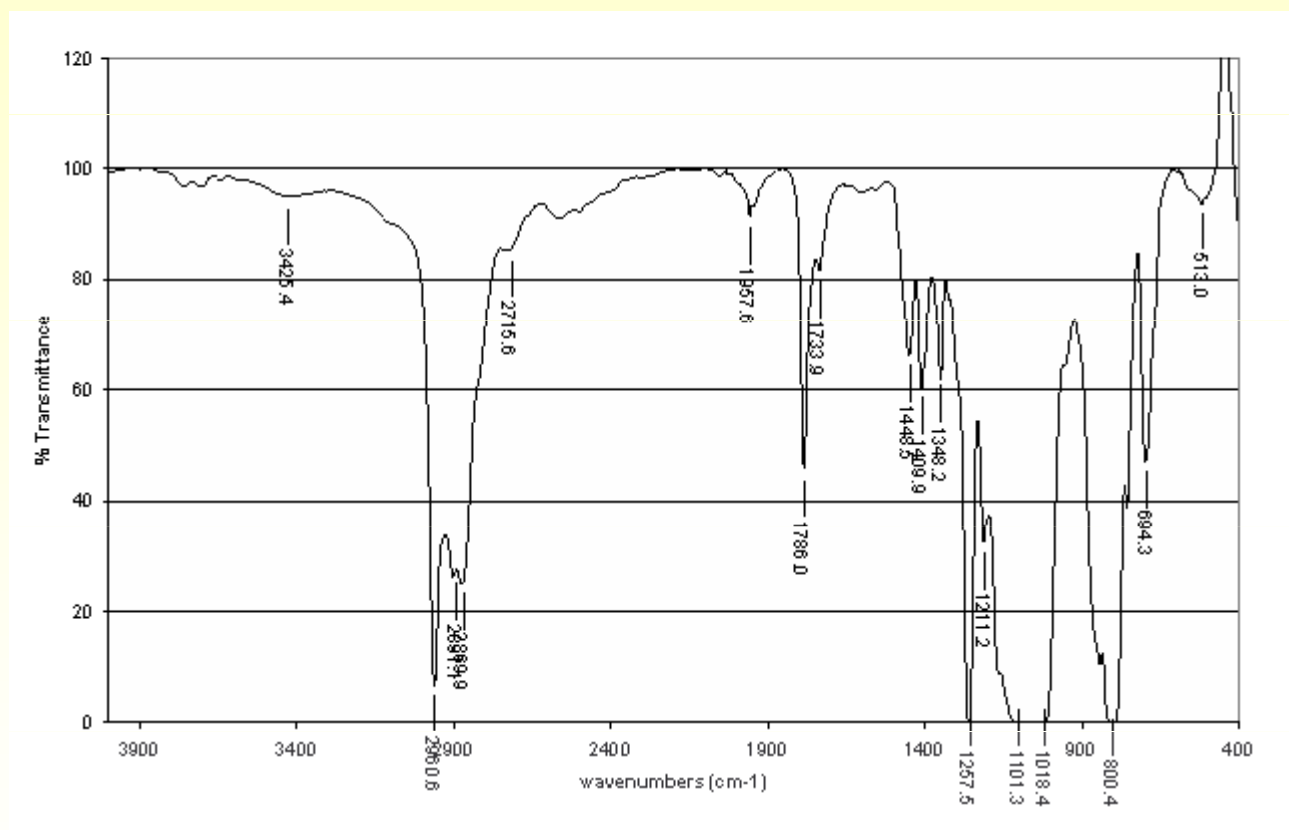
$^1\text{H}$  NMR of PEG-AA-PDMS

# RESULTS ...Task I : Synthesis of APGC



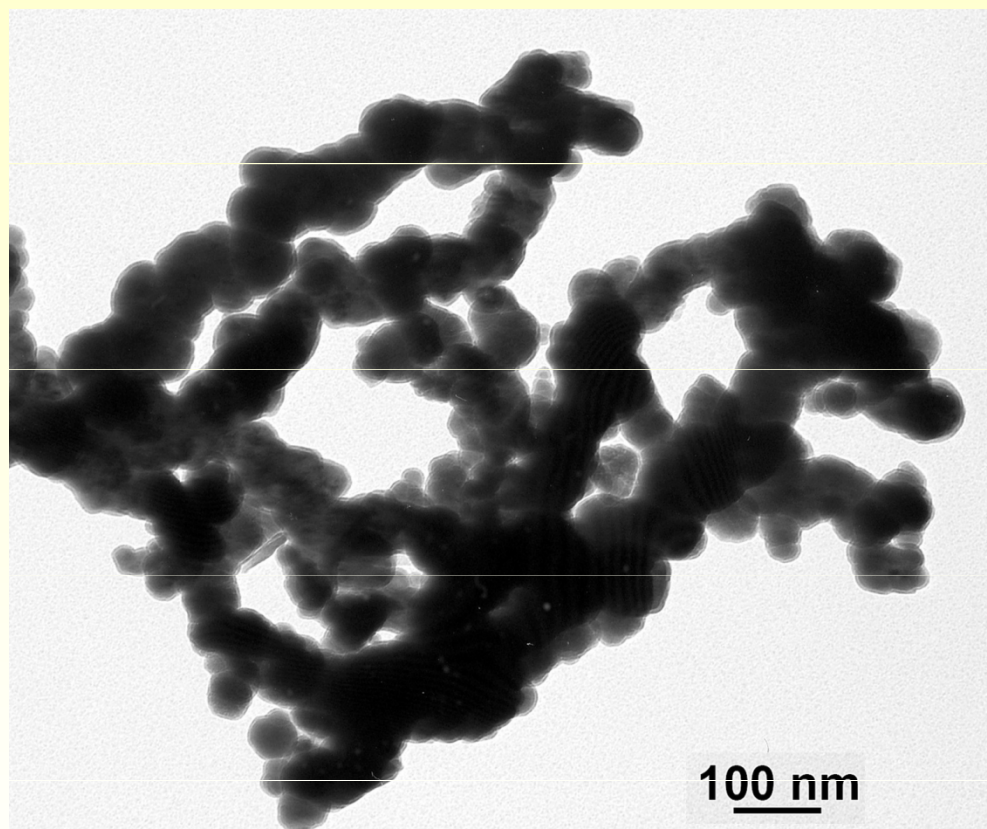
$^{13}\text{C}$  NMR of PEG-AA-PDMS

# RESULTS ...Task I : Synthesis of APGC



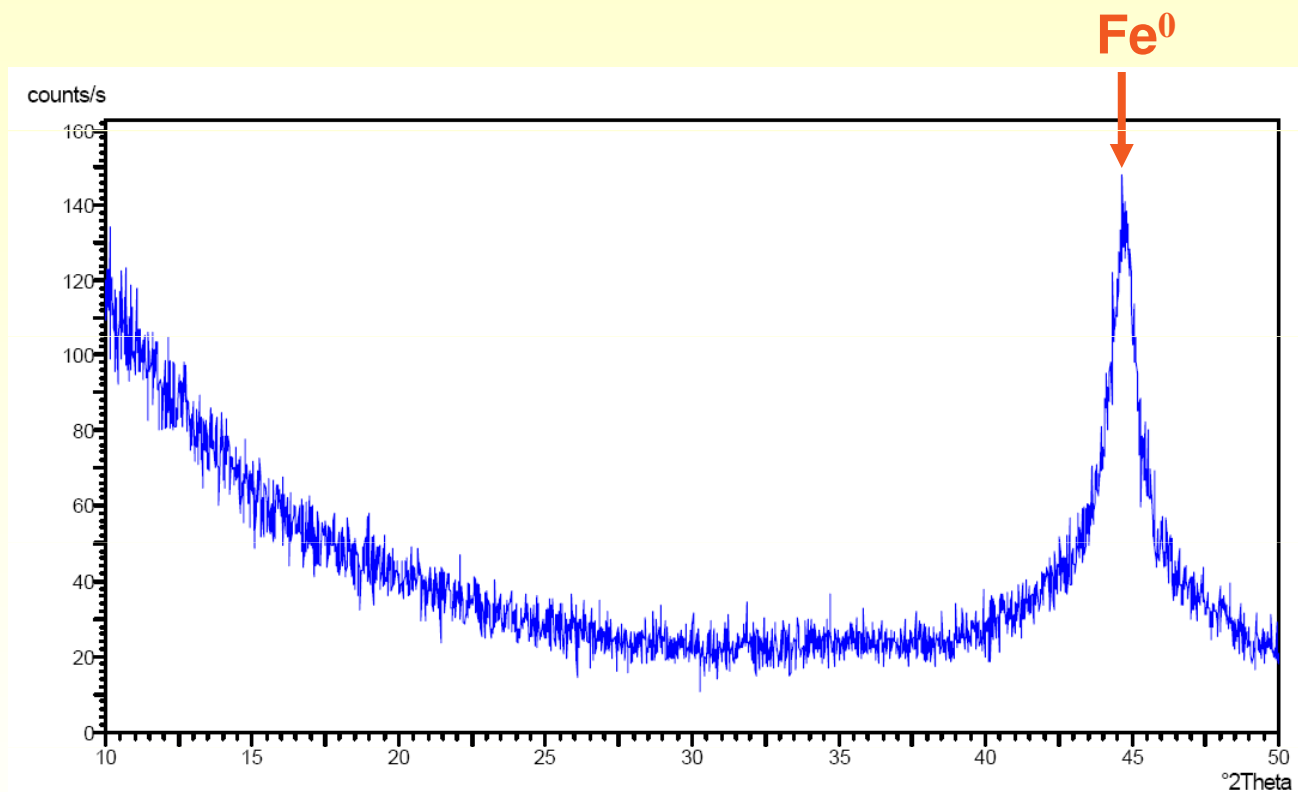
FTIR spectra of PEG-AA-PDMS

## RESULT...TASK II : Synthesis of nZVI



TEM image of  $\text{Fe}^0$  nanoparticles

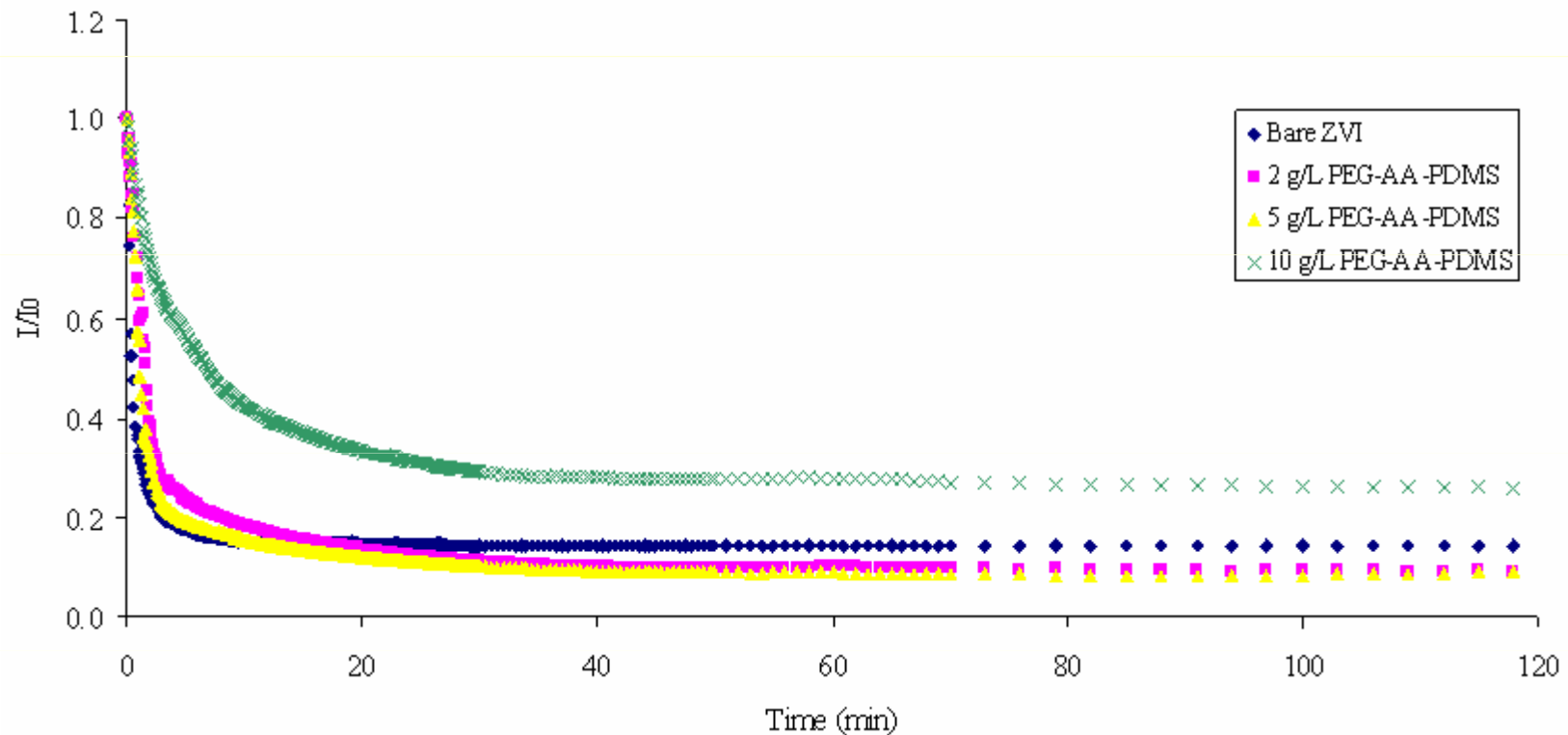
# RESULT...TASK II : Synthesis of nZVI



XRD spectra of Fe<sup>0</sup> nanoparticles



# TASK III: Colloidal Stability Study



Use of 10g/L of APGC gives the best colloidal stability to nZVI

# SUMMARY

- Task I : Synthesis of APGC
  - This task is already completed
- Task II: Synthesis of Fe<sup>0</sup> Nanoparticles
  - This task is already completed
- Task III: Colloidal Study
  - Working on various APGCs and modified nZVIs
- Task IV: Kinetic Studies
  - Batch studies will be conducted after Task III
- Task V: TCE by-products and characterization of Fe<sup>0</sup>
  - This part will be conducted after Task IV

# ACKNOWLEDGMENT

- Achintya Bezbaruah (Advisor),  
Department of Civil Engineering, NDSU
- Bret Chisholm (Co-advisor),  
Center for Nanoscale Science and Engineering (CNSE), NDSU
- Laura Jarabek, Eric Jarabek, Shane Stafslie, David  
Christianson, Bret Mayo, Alexander Kugel  
CNSE, NDSU
- Jay Thompson, Graduate Student, Civil Engineering, NDSU
- My friends in Environmental Engineering Laboratory  
Department of Civil Engineering, NDSU

---

# THANK YOU